

Green synthesis of silver nanoparticles from plant sources and evaluation of their antimicrobial activity

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ABSTRACT

Nanoparticles are used widely due to its small size, orientation, physical properties, which are reportedly shown to change the performance of any other material which is in contact with these particles. Silver nanoparticles have received substantial attention in the field of antimicrobial research due to the antimicrobial activity of silver. These particles can be synthesized easily by different chemical, physical, and biological approaches. The biological approach is the most emerging approach of preparation, as this method is easier than the other methods, ecofriendly, non toxic and less time consuming. In the present study green synthesis of silver nanoparticles was carried out using the plant extracts of Neem (Azadirachta indica), Tea (Camellia sinensis), Peels from kitchen waste and the mixture of all the three extracts. Silver was of a particular interest for this process due to its evocative physical and chemical properties. The reduction process of Ag⁺ to Ag⁰ was observed by formation of brown colour within 24 hours of incubation period and synthesized SNPs showed an absorption peak at around 450 nm in the UV-visible spectrum. The reaction was followed by the characterization of the silver nanoparticles using UV-Vis and FTIR. The antimicrobial activity was tested using agar well method with Gram positive and Gram negative microorganism. It was observed that the antimicrobial activity was higher against Gram negative bacteria. Local plants such as tea leaves and waste is an alternative to a safer, more ecofriendly alternatives of synthesizing silver nanoparticles. This environment friendly, method provides simple, easy and cost effective faster synthesis of nanoparticles than chemical method and can be used in several areas such as catalysis, medical application etc.

Keywords: Silver nanoparticles, UV-Visible Spectroscopy, FTIR, Antimicrobial activity

I. INTRODUCTION

Nanotechnology is the branch of technology that deals with dimensions and tolerances of less than 100 nanometers, especially the manipulation of individual atoms and molecules. They exhibit novel and significantly improved physical, chemical and biological properties, phenomena and processes because of their size [1]. Nanoparticle may or maynot exhibit size-related properties that differ significantly from those observed in fine particles or bulk materials [2]. Some of the applications of nanoparticles to biology or medicine are in tissue engineering, cancer therapy, multicolour optical coding for biological assays, manipulation of cells and biomolecules, protein detection, drug and gene delivery, bio detection of pathogens, probing of DNA structure, tumour destruction via heating (hyperthermia), MRI contrast enhancement and phagokinetic studies.

Among the various metal nanoparticles, silver (Ag) nanoparticle has received substantial attention in the





field of antimicrobial research. Silver has long been recognized as having an inhibitory towards many effect bacterial strains and microorganisms [3]. Synthesising silver nanoparticles by plants is a major advantage that they are easily available, safe and nontoxic in most cases, have a broad variety of metabolites that can aid in reduction of silver ions and are quicker than microbes in the synthesis Biosynthesis of nanoparticles is a kind of bottom up approach, where the main mechanism of action is redox reaction. The need for biosynthesis of nanoparticles arises from the fact that the physical and chemical processes became costly and the use of hazardous chemicals in various steps [4]. Phytochemical mediated metal nanoparticle syntheses effective. are economical and environmental friendly. Plants are considered as biosynthetic laboratories of wide spectrum of phytochemicals such as phenolics, alkaloids and flavonoids. These phytochemicals are expected to self assemble and cap the metal nanoparticles formed in their presence and thereby induces some shape control during metal ion reduction [5]. Metallic nanoparticles obtained by biosynthesis employing plant extracts have been reviewed and reported by different authors [6,7].

In this work we have synthesized and characterized the silver nanoparticles from plant sources like leaves of Tea, Neem, Peels from kitchen waste and also from the combination of all three extracts and have checked the antibacterial activity of the synthesized silver nanoparticles

II. MATERIALS AND METHODS

Test chemicals

AgNO₃were purchased from SD Fine Company, India and used without further purification. The purity was atleast 99.5%. The solutions for the metal salts were prepared in deionised water. The media for bacterial growth was obtained from Himedia, India.

Sample Collection:

The sample sources for synthesis of nanoparticles were dried Tea leaves (Camellia sinensis)from local market,fresh Neem leaves (Azadirachtaindica)from Ambivili area and Peels of mango, sweet potato, drumstick, tomato, carrot, pumpkin, snake gourd collected from normal kitchen waste.

Preparation of extract:

Samples used were fresh leaves of Neem (Azadirachtaindica), dried leaves of Tea (Camellia sinensis) and Peels of mango, sweet potato, drumstick, tomato, carrot, pumpkin, snake gourd.The solvent used was deionized water.

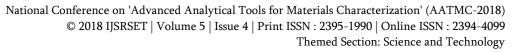
The samples were weighed 1g each and added to 50ml of deionized water in a flask. These flasks were kept in boiling water bath for 15min with inverted funnel. After cooling, the solution was filtered using Whatmanfilter paper no. 1. 2ml each of the filtrates were mixed in a conical flask and used as the fourth sample. The samples were named Tea, Peel, Neem and Mix [8].

Synthesis of Silver nanoparticles:

1 ml of freshly prepared 5mM Silver nitrate solution was added to 5ml of the sample solutions respectively. This mixture was kept in boiling water bath for 10min until visible colour change (brown colour) was observed. The mixtures were kept in incubator at 37°C for 24hrs [8].

UV-Vis spectral analysis

The reduction of pure Ag⁺ ions was monitored by using UV-Vis spectrum of reaction medium after 24







hrs. Sample was prepared by diluting the solution in deionized water in the ratio 1:10. Deionized water was used as blank.The analysis was carried out on a UV – Vis spectrophotometer (Jasco v-

630 Spectrophotometer) from range 350-600nm.

FTIR analysis of silver nanoparticles

The chemical composition of synthesized silver nanoparticles was studied by using FTIR spectrometer (JASCO FT/IR 4100 type A). The liquid samples were mixed with KBr and pellets were made. The solution were characterized in the range 4000-650 cm⁻¹

Study of antimicrobial activity

The antibacterial activity of the synthesized silver nanoparticles was checked using Agar Well Diffusion method. 1 ml of the 24 hrs old grown bacterial culture was seed inoculated in molten sterile Nutrient agar butt and poured into petriplates. All the four samples (Neem, Tea, Veg peels, Mix) were tested against the five cultures - Staphylococcus aureus, Escherichia coli, Klebsiella pneumoniae, Corynebacterium diphtheriae and Pseudomonas aeruginos.

After the incubation period, the plates were examined for zone of inhibition. The diameter of the zones of clearance was measured and the mean value was calculated.

III. RESULT AND DISCUSSION

The mixture showed colour change to brown colour at 37°C, after addition of plant extract to aqueous solution of AgNO₃. The colour is characteristic of the Surface Plasmon Resonance (SPR) of silver nanoparticles. The reduction of silver ion to silver nanoparticle was reflected in spectral data obtained by using a UV-Vis spectrophotometer. Absorbance was found to be in range of 350-450nm which is specific for silver nanoparticles (Table 1, Figure 1) [9].

Table 1	1: Abs	orbance	of the	AgNPs
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Silver	Maximum	Absorbance		
nanoparticles	wavelength			
Neem	443.8nm	0.477		
Green tea	369.6nm	0.997		
Vegetable Peels	431.6nm	0.556		
Mixture of all	447nm	0.791		
three				

Neem is found to exclude addition of external stabilizing agent during synthesis of silver nanoparticles and to offer synergistic effects to enhance the antimicrobial properties of the synthesized silver nanoparticles [10].

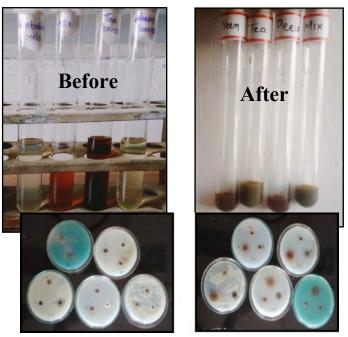


Figure 1. Colour change during synthesis of AgNPs before and after



The UV–Vis spectra confirmed the formation of Ag nanoparticles as the colour change occurred to reddish brown and maximum absorbance was found at 443.8nm for Neem AgNPs which lies in the range of earlier reported biosynthesized silver nanoparticles, 350-550nm [10] and 350-450nm [9] Other studies stated its range to be mostly around and at [11,12].

The Green tea leaves which were used for synthesis of silver nanoparticles showed peak at 369nm [13,14] The peels from kitchen waste also proved to be a good source for synthesis of nanoparticles. The UV-Vis spectral analysis showed its maximum absorbance at 431.6nm which was found to lie near the previously observed peak of 430nm with similar peels [15]

FTIR Analysis

FT-IR was carried to determine the functional groups responsible for reduction present in the plant sample and the synthesized nanoparticles. The FT-IR spectrum obtained for extracts displays a number of absorption peaks, reflecting its complex nature. Strong absorption peaks at 3263 to 3331 cm⁻¹ result from stretching of the –NH band of amino groups or is indicative of bonded –OH hydroxyl group. The absorption peaks at about 2929.87 cm⁻¹ could be assigned to stretching vibrations of -CH2 and -CH3 functional groups. The intense band at 1585 cm⁻¹

The characterization of Neem silver nanoparticles by FTIR showed the presence of alcohols and phenols, alkenes, aromatic group and ether which matched with the previously reported groups alkenes, alcohols and phenols with an additional primary amine group at 1634.90cm⁻¹ [11]. In Peel AgNPs presence of aromatic groups, ketone groups and alcohols and

could be assigned to the aromatic C=C bending. FT-IR study indicates that the hydroxyl (-OH), and amine (N-H) groups in the extracts are mainly involved in reduction of Ag⁺ ions to Ag⁰ nanoparticles. For the plant extract, the FTIR results confirmed the presence the presence of -NH, -OH, C=C, and -CH group, which indicates that the plant extract containing the hydroxyl and amine group substituted flavonoids. It is observed that flavonoids act as a reducing agent, which reduces Ag⁺ to Ag⁰ and the amino group as stabilizing agents in the green synthesis of silver Nanoparticles. The FT-IR spectroscopic studies have also showed that the leaf extracts not only acts as reducing agent but as a stabilizer also, which prevents the agglomeration of AgNPs. The carbonyl group of amino acid residues has a strong binding ability with metal, suggesting the formation of a layer covering silver nanoparticles which act as a stabilizing agent to prevent agglomeration in the aqueous medium [16]. The observed intense band were compared with standard values of identify the functional groups.

The groups assisting reduction were found to be in the range of 1400-1600, 1640-1690, 2850-3000 and 1000-1300cm⁻¹. Thus the reduction of plant extracts by AgNO₃ was confirmed by FTIR analysis and the flavonoids present were found to be responsible for reduction.

phenols were observed indicating presence of flavonoids and other phytochemicals which were previously reported with an additional amine group at 3403.99cm⁻¹ [15] The antimicrobial activity of the silver nanoparticles was studies against five different organisms viz. Gram positive organisms (Staphylococcus aureus and Corynebacterium



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diphtheria)and Gram negative organisms (Escherichia coli, Klebsiella pneumoniae and Pseudomonas aeruginosa)(Figure 2, Table 2).

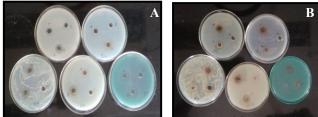


Figure 2. Antimicrobial activity of AgNP's prepared from [A] neem leaves, [B] tea leaves, [C] vegetable peels and [D] mixture of all.

Table 2. Zone of inhibition for the AgNPs

Organisms	Zone of inhibition(mm)					
	Neem	Tea	Peels	Mixture	Control A	Control B
	AgNP	AgNP	AgNP		(1mM AgNO3)	(Plant extract)
Escherichia coli	13	14	12	13	_	-
Klebsiella pneumonia	12	14	15	16	-	-
Pseudomonous aeruginosa	11	11	13	12	-	-
Staphylococcus aureus	11	11	13	14	_	-
Corynebacterium	13	18	19	19	_	-
diphtheriae						

The figures show clear zone of inhibition in the well treated with silver nanoparticles whereas the well treated with only plant extract and silver nitrate did not show any inhibition. It was observed that the antibacterial activity of Mix AgNPs was more compared to Neem, Tea and Peels AgNPs against Staphylococcus aureus, Escherichia coli, Klebsiella pneumoniae and Corynebacterium diphtheria. All the four nanoparticles showed maximum activity against Corynebacterium diphtheria and minimum against Pseudomonas aeruginosa.

The Neem nanoparticles showed inhibitory action against Staphylococcus aureus, Escherichia coli and Pseudomonas aeruginosa which was previously reported [10]. In addition, antibacterial action against Klebsiella pneumoniae and Corynebacterium diphtheria was also observed. Considerably good antibacterial activity of Peel nanoparticles was observed against Escherichia coli and Klebsiella pneumoniae as earlier stated [15].

IV. CONCLUSION

The research work suggests that plants extract made from Neem, Tea and Peels from Kitchen waste and mixture of all capable of producing silver nanoparticles extracellular. The use of plant in synthesis of nanoparticles is advantageous in many ways in terms of production, economic viability and safety. The samples were environmental friendly, simple and efficient route for synthesis of benign characterization nanoparticles. The of silver was carried out using UV-Vis nanoparticles spectrometer and FTIR. The characterization by FTIR



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showed the presence of aromatic groups, ketone groups and alcohols and phenols indicating presence of various phytochemicals including flavonoids which aid in reduction of silver. Plant extract can be used efficiently in synthesis of silver nanoparticles as greener route. Control over physical dimensions seems to be very easy with the use of plants. Nanoparticles produced from plants have various applications. The above silver nanoparticles revealed to have effective an antimicrobial property against Staphylococcus aureus, diptheria, Corynebacterium Escherichia coli. Klebsiella pneumonia and Pseudomonas aeruginosa. The present study emphasizes the use local plants and waste part for synthesis of silver nanoparticles with potent antibacterial effect. This green synthesis method has many advantages such as eco-friendly, cost effective and easily scaled up to large scale synthesis. The further characterization of the silver nanoparticles can be done by SEM analysis preferably to find its size and shape. Various parameters including pH, temperature and concentration of AgNO₃ can be altered to obtain better results. Further, the checking of its antioxidant property may give new applications for the use of silver nanoparticles.

V. REFERENCES

- Kannan N., Shekhawat M.S., Ravindran C.P. and Manokari M. (2014). Preparation of silver nanoparticles using leaf and fruit extracts of Morindacoreia Buck., Ham. –A green approach. Journal of Scientific and Innovative Research. 3(3): 315-318.
- Banerjee P.,Satapathy M., Mukhopahayay A. and Das P. (2014). Leaf extract mediated green synthesis of silver nanoparticles from widely available Indian plants: synthesis,

characterization, antimicrobial property and toxicity analysis. Bioresources and Bioprocessing. 1:3.

- Sriram T.and Pandidurai V. (2014). Synthesis of silver nanoparticles from leaf extract of Psidiumguajava and its antibacterial activity against pathogens. Inernational Journal of Current Microbiology and Applied Sciences. 3(3):146-152.
- 4. Ramya M, Sylvia Subapriya (2012) Green synthesis of silver nanoparticles. Int J Pharm Med Bio Sc 1: 54-61.
- Ahmad N., Sharma S., Alam M.K., Singh V.N., Shamsi S.F.(2010) Rapid synthesis of silver nanoparticles using dried medicinal plant of basil. Colloids Surf B 81: 81-6.
- Bali R., Razak N., Lumb A., Harris AT. (2006) The synthesis of metallic nanoparticles inside live plants. Proce Int Conference Nanosci Nanotechnol (ICONN '06) 224-7.
- Victor S.M., Alfredo R.V.N. (2012) Green Synthesis of Noble Metal (Au, Ag, Pt) Nanoparticles, Assisted by Plant-Extracts. Noble Metals: InTech 392-408
- Sankar R, Karthik A, Prabu A, Karthik S, Shivashangari KS, Ravikumar V. Origanum vulgare mediated biosynthesis of silver nanoparticles for its antibacterial and anticancer activity. Colloids Surf B Biointerfaces. 2013;108:80-4.
- 9. Bose D. and Chatterjee S. (2016). Biogenic synthesis of silver nanoparticles using guava (Psidiumguajava) leaf extract and its antibacterial activity against Pseudomonas aeruginosa. Applied Nanoscience. 6:895–901.
- Namratha N.and Monica P.V. (2013). Synthesis of silver Nanoparticles using Azadirachtaindica(Neem) extract and usage in water purification. Asian J. Pharm. Tech. 3(4):170-174.
- 11. Agrawal P., Mehta K., Vashisth P., Sudarshan, Bhat P., VishnuB. V. G. (2014). Green Synthesis of Silver Nanoparticles and Their Application in



Dental Filling Material. International Journal of Innovative Research in Science, Engineering and Technology. 3(6):13038-13052.

- Gavhane A. J., Padmanabhan P., Kamble S. P. and Jangle S. N. (2012). Synthesis of silver nanoparticles using extract of Neem leaf and Triphala and evaluation of their antimicrobial activities. International Journal of Pharma and Bio Sciences. 3(3):88-100.
- Loo Y. Y., Chieng B. W., Nishibuchi M. and Radu S. (2012). Synthesis of silver nanoparticles by using tea leaf extract from Camellia Sinensis. International Journal of Nanomedicine. 7:4263-4267.
- 14. Shet A. R., Tantri S. and Bennal A. (2016). Economical biosynthesis of silver nanoparticles using fruit waste. Journal of Chemical and Pharmaceutical Sciences. 9(3):2306-2311.
- Sharma K., Kaushik S., Jyoti A. (2016). Green Synthesis of Silver Nanoparticles by Using Waste Vegetable Peel and its Antibacterial Activities. Journal of Pharmaceutical Science & Research. 8(5):313-316.
- 16. Awwad A.M., Salem N.M., Abdeen A.O. (2013) Green synthesis of silver nanoparticles using carob leaf extract and its antibacterial activity. Int J Ind Chem 4: 2